PCT

PCT

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicant's or agent's file r PFC 1465 PCT		HER ACTION		ation of Transmittal of International Examination Report (Form PCT/IPEA/416)
International application No	o. International fi	iling date (day/month	/year)	Priority date (day/month/year)
PCT/GB00/02387	20/06/2000)		24/06/1999
International Patent Classi C01B3/40	fication (IPC) or national classificati	ion and IPC		
Applicant				
JOHNSON MATTHE	Y PUBLIC LIMITED COMPA	ANY et al.		
	oreliminary examination report had the applicant according to Ar		by this Inte	rnational Preliminary Examining Authority
2. This REPORT cons	sists of a total of 4 sheets, incl	uding this cover sh	ieet.	
been amended		ort and/or sheets co	ontaining red	n, claims and/or drawings which have ctifications made before this Authority e PCT).
These annexes cor	nsist of a total of 5 sheets.			
3. This report contains	s indications relating to the follo	owing items:		
I ⊠ Basis o	of the report			
II □ Priority	•			
III 🗆 Non-es	stablishment of opinion with reg	ard to novelty, inv	entive step a	and industrial applicability
IV 🗆 Lack of	f unity of invention			
	ned statement under Article 35(as and explanations suporting s		ovelty, inve	ntive step or industrial applicability;
_	documents cited			•
VII □ Certain	defects in the international app	plication		
VIII □ Certain	observations on the internation	nal application		
			-	<
Date of submission of the o	temand	Date of c	ompletion of t	his report
16/01/2001		18.07.20	01	
Name and mailing address preliminary examining auth		Authorize	ed officer	STONE ON THE VIEW OF
European Pat D-80298 Mun Tel. +49 89 23		Mayne,	J	
Fax: +49 89 2	399 - 4465	Telephon	ie No. +49 89	2399 8572

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/GB00/02387

I.	Bas	sis fth rep rt				
1.	the and	receiving Office in	ments of the international applic response to an invitation under to this report since they do not c	Article 14 are	referred to in this repo	ort as "originally filed"
	1,4,	,6,7	as originally filed			
	2,3,	,5	as received on	05/07/2001	with letter of	02/07/2001
	Cla	ims, No.:				
	1-8		as-received on	05/07/2001	with letter of	.02/07/2001
	Dra	wings, sheets:				
	1/2,	,2/2	as originally filed			
2.			guage, all the elements marked international application was file			
	The	se elements were	available or furnished to this Au	thority in the fo	ollowing language: ,	which is:
			translation furnished for the pur		•	nder Rule 23.1(b)).
			ublication of the international ap	•	,	
		the language of a 55.2 and/or 55.3).	translation furnished for the pur	poses of inter	national preliminary ex	camination (under Rule
3.			cleotide and/or amino acid sec ry examination was carried out o			l application, the
		contained in the ir	nternational application in written	form.		
		filed together with	the international application in o	computer read	lable form.	
		furnished subsequ	uently to this Authority in written	form.		
		furnished subsequ	uently to this Authority in comput	ter readable fo	orm.	

☐ The statement that the subsequently furnished written sequence listing does not go beyond the disclosure in

☐ The statement that the information recorded in computer readable form is identical to the written sequence

4. The amendments have resulted in the cancellation of:

the international application as filed has been furnished.

listing has been furnished.

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/GB00/02387

		the description,	pages:		
		the claims,	Nos.:		
		the drawings,	sheets:		
5.					some of) the amendments had not been made, since they have been as filed (Rule 70.2(c)):
		(Any replacement shoreport.)	eet contail	ning such	h amendments must be referred to under item 1 and annexed to this
6.	Add	itional observations, if	necessar	y:	
٧.		soned statement un tions and explanatio			vith regard to novelty, inventive step or industrial applicability; ch statement
1.	Stat	ement			
	Nov	elty (N)	Yes: No:	Claims Claims	
	Inve	ntive step (IS)	Yes: No:	Claims Claims	· ·
	Indu	strial applicability (IA)	Yes: No:	Claims Claims	

Form PCT/IPEA/409 (Boxes I-VIII, Sheet 2) (July 1998)

2. Citations and explanations see separate sheet

EXAMINATION REPORT - SEPARATE SHEET

R It m V

Reasoned statement under Rule 66.2(a)(ii) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

Reference is made to the following document: D1 EP-A-0884271

Article 33(2) and (3) PCT

None of the cited prior art documents discloses all the features of the independent claims 1 and 2. Claims 1-8 are therefore novel and fulfill the requirements of Article 33(2) PCT.

D1 is concerned with a methanol reformer which contains a catalyst and is used to supply H₂ to a fuel cell (e.g. claim 1 of D1 and abstract). D1 discusses various ways to reactivate the catalyst whilst it is being used which include:

reducing the fuel load; raising the temperature; increasing the air λ value to increase the O₂ excess

(see col. 3, I. 10-26, claim 1, col. 5, I. 1st paragraph and col. 6, I. 33-36).

For the catalyst to be regenerated the reforming reaction is interrupted according to D1. claim 1. D1 does not specify that fuel, air and steam continue to be passed through a reforming catalyst during the regeneration phase.

In the examples of the application in which fuel, air and steam continue to be passed through a reforming catalyst whilst regeneration occurs, either using air pulses or raised temperature, the Applicant has shown that he can maintain a hydrogen concentration of above 25% throughout the operation (example 2, Fig. 1b, example 3). This effect can also be achieved by increasing the steam feed rate (example 5) or by inhibiting deactivation of the catalyst either using an additive (example 4) or extra air (example 6).

Modulating the fuel feed-rate would also alter the oxygen to carbon ratio and be expected to have a similar effect as modifying the air feed-rate.

Since the methods of claims 1 and 2 are not derivable directly from D1 and the technical effects thereby achieved are not foreseeable from D1 an inventive step is accorded.

Claims 1-8 fulfill the requirements of Article 33(3) PCT.

phod 34

Because the performance of the reformer has a direct effect on the power output from the fuel cell, deactivation of the catalyst cannot be allowed to proceed unchecked. However the process of power generation cannot be stopped to allow regeneration of the reformer, when its performance falls below an acceptable level. The regeneration of such catalytic reactors whilst they are in use is the problem this invention sets out to solve.

In many instances, the performance of the catalyst declines while it is being used. Although the rate of decline can usually be reduced by changing the operating conditions, it is often accompanied by a drop in hydrogen and power output. One frequently practised method for restoring the performance of a fuel processor is to replace the catalyst periodically. Another method involves regenerating the catalyst while it remains in the processor, using a prescribed regeneration procedure (eg see JR Rostrup-Nielsen in Catalysis Today, Vol 37, 1997, p 225-232). However, both these require that the process of hydrogen-generation is interrupted in order to restore performance.

Thus in a first aspect, the present invention provides a method for regenerating a catalytic fuel processor, while it is being used to supply hydrogen to a fuel cell, comprising any one or more of the steps of:

20

25

5

10

15

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,
- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,
- continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.
- continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

30

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

The reforming catalyst may be one component of the catalytic fuel processor, or it may be the only component. Where loss of performance is being caused by a particular component in a complex fuel mixture, the regeneration method can be targeted at the specific deactivating effect.

5

Thus in a second aspect, the present invention provides a method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:

10

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,
- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,

15

- continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel,
- continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

20

25

30

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

Again, the reforming catalyst may be one component or the only component of the catalytic fuel processor. In a preferred embodiment of the invention water is temporarily added to the feed. ('Feed' is a term used to describe the reactant mixture supplied to the fuel processor.) The water will be rapidly converted into steam in the catalyst. The water (steam) to carbon ratio in the feed may readily be adjusted in order to permit targeting of the regeneration method to a specific deactivating event. Adjusted for the purposes of this invention means increased or decreased. The oxygen to carbon ratio may also be adjusted for the same purpose, by adding or changing the feed-rate of air, or by changing the feed-rate of the fuel.

Figure 1b shows the effect of air pulses on the composition of dry reformate during reforming of dodecane (as described in Example 2).

5

In both cases % in reformate is on the vertical axis and time/hours is on the horizontal axis.

5 ◆ represents Hydrogen, ■ represents carbon monoxide, and ▲ represents carbon monoxide.

Figure 2a shows the composition of dry reformate as a function of time during the reforming of toluene under the two conditions described in example 5.

Figure 2b shows the effect of increasing the air feed-rate on the composition of dry reformate, during reforming of toluene (as described in example 6).

In both cases the left hand side of the vertical axis represents % in reformate and the right hand side of the vertical axis represents temperature in °C. The horizontal axis represents time/hours; ◆ represents hydrogen, ■ represents carbon dioxide,

 \triangle represents carbon monoxide and \mathbf{x} represents temperature.

15

20

25

30

It will be appreciated that many variations can be made to the invention herein described without departing from the present inventive concept.

Example 1 (Aliphatic fuel; no regeneration)

A bed (0.2 g) of reforming catalyst was packed into a tubular quartz reactor, which was positioned at the centre of a furnace. A mixture of dodecane vapour (produced by vaporising the liquid at a rate of 4 cm³ hour-¹), air (200 cm³ min-¹) and steam (produced by vaporising water at a rate of 4 cm³ hour-¹) was passed through the catalyst bed, which was heated by the furnace. The furnace temperature was maintained at 500°C. The product stream (ie the reformate) passed through a drier before entering a gas chromotagraph, which was used to analyse for hydrogen. The initial concentration of hydrogen in the dry reformate was just above 25%, but declined to 20% within an hour, and to 15% within 4 hours. See Figure 1a for results.

CLAIMS

A method for regenerating a catalytic fuel processor, while it is being used to 1. supply hydrogen to a fuel cell, comprising any one or more of the steps of:

5

continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,

10

continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate.

continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.

continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

15

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

20

2. A method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:

25

continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,

continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,

continuing to pass air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.

30

continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

9

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

- 3. A method according to either claim 1 or claim 2, whereby water is temporarily added to the fuel.
 - 4. A method according to claims 1 or 2 in which air is temporarily added to the feed.
- 10 5. A method according to claims 1 or 2 in which an additive is added to the feed.
 - 6. A method according to claim 5 in which the additive is an oxygenate.
- 7. A method according to claim 6 in which the oxygenate is MTBE (methyl-tert-butylether).
 - 8. A method according to claims 1 or 2 in which the catalyst bed temperature is raised temporarily by an external heat source.
- 9. A method according to claims 1 or 2 in which the temperature of one or more of the reactant feeds is raised temporarily.

Applicants or agent's file reference PEC 1465 PCT

GRANDE BRETAGNE

International application No. PCT/GB09/02387 International filing date (day/month/year) 20/06/2000

Priority date (day/month/year) 24/06/1999

Applicant

JOHNSON MATTHEX PUBLIC LIMITED COMPANY of al.

- 11. The applicant is hereby notified that this International Preliminary Examining Authority transmits herewith the intermational preliminary examination report and its annexes, if any, established on the international application.
- 2. A capy of the report and its annexes, if any, is being transmitted to the International Bureau for communication to all the elected Offices.
- Where required by any of the elected Offices, the International Bureau will prepare an English translation of the report touthout of any annexes) and will transmit such translation to those Offices.

4 REMINDER

The applicant hust enter the national phase before each elected Office by performing certain acts (filing translations and paying national fees) within 30 months from the priority date (or later in some Offices) (Article 39(1)) (see also the reminder sent by the international Bureau with Form PCT/IB/301).

Where a translation of the international application must be furnished to an elected Office, that translation must contain a translation of any annexes to the international preliminary examination report. It is the applicant's responsibility to prepare and furnish such translation directly to each elected Office concerned.

For further details on the applicable time limits and requirements of the elected Offices, see Volume II of the PCT Applicants Guide.

Name and grading address of the IPEA

Authorized officer

Koutsoftas, P

European Patent Office

D 80298 Muldel

E 49 89 239 - 0 Tx: 523656 epmu d

ax: 49 89 239 - 4465

Tel.+49 89 2399-7273

Form PCI HEA/416 (July 1992)

U DILE SOLL

PATENT COOPERATION THEATY PCT

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicantsion agants the sevence			1
PFC 1465 PCT	FOR FURTHER ACTION	See Notification of Transmittal of Prefiminary Examination Report	International (Form PCT/IPEA/416)
Por GE00/02367	International filing date (day/month 20/06/2000	Priority date (day/m 24/06/1999	onth/year)
International Patent Classification (IPC) or na	tional classification and IPC		
C6183/4b		•	
Applicant			
JOHNSON MATTHEY PUBLIC LIM	TED COMPANY et al.		
1. This International preliminary exam	ination report has been prepared	by this International Preliminar	v Examining Authority
and is transmitted to the applicant a	coording to Article 36.	_	
2. This REPORT densists of a total of	4 sheets, including this cover sh	reet.	
Tijis report is also accompanier	I by ANNEXES, i.e. sheets of the is for this report and/or sheets co	e description, claims and/or dra	wings which have
(see Rule 70.18 and Section 60	7 of the Administrative instruction	ins under the PCT).	(ore this Authority
			:
These annexes consist of a total of	5 sheets.		1
			:
3.1 This report contains indications rela	ting to the following items:		\$ }
			1
⊠ Basis of the report			
Pribrity			
	ninion with regard to novelty, inve	entive step and industrial applic	ability
citations and explanatio	der Article 35(2) with regard to no ns suporting such statement	ovelty, inventive step or industr	al applicability;
VI 🔲 Certain documents cite			i !
Vii 🔲 Censin defects in the In	ternational application		
VIII Certal hobservations on	the international application		i ,
: 44 1 3 3 1			1
Date of submission of the demand	Date of cy	empletion of this report	1
		institution of any report	
16/01/2001	18.07.200	ห	
: #### <u> </u>			
Name and mailing address of the International pseliminant examining authority:	Authorize	d officer	A STATE OF THE PARTY OF THE PAR
European Patent Office	į		
D-80298 Munity	Mayne,	J	(4.99)
Tel -49 86 2399 - 0 Tx: 523656	Talanhan	e No. +49 89 2399 8572	A Solar West
1970 H. BOND TO 1971 1971 1			<u> </u>

Form (January 1994)

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/GB00/02387

	Ba	si s	oj t	he i	epc	1							
	Ν'n	hré	gar	dito	ήe	ee	ments of t	ne internati	onal applica	ation (Replac	ement sheets	which have	e been fumished to
: } {	116	л е с	ΕĮVI	ny (ŊЩÇ	e m	response i	to an Invita:	tion under l	Article 14 are	referred to in	thic monet	Se "neiginally filed"
	وا	ći,	ptic) , _[æg		io iriis repu	it snice uie	y do not do	xitain amend	ments (Rules	/U.16 and	70.17)) : :i
: :		6.7	;	٠:			ao ociaina	المساكة مواك					ባ የ !
1.7	1	ve č	3	i	٠	- [;	as origins	m y me o	,				i
1	3,	5	:	;	!	ji H	as receive	ed on		05/07/2001	with letter of	0	2/07/2001
		41,	Ì				ı						• •
: (lai	ms	No) ‡	į ·	:							! !
į	-8	,	,		į :		as receive	ed on		05/07/2001	with letter of	Δ.	 2/07/2001
4)		•	•	:	į.					00,0772001	with lotter or	V _i	2107/2001 !
	ra	wine	as.	she	ets		•						}
: ij	الم		-										· {
1	/2,2	20,22	1				as original	lly filed					1
]] ,			:	•	١,	ļi.							
	;		ļ		ነ 1								
, v	lijh	reg	ard	to t	he l	ang	uage, all ti	ne element	s marked a	bove were a	vailable or fur	ished to th	is Authority in the
i ja	rigi	u ag	e in	wh	ich 1	ha i	internationa	al application	on was filed	l, unless othe	rwise indicate	d under thi	s item.
-	i nes	eje	em	ent	we	ne a	Ivaliable or	furnished t	to this Auth	ority in the fo	llowing langua	iae: whi	ch is:
- #		· ,	i i	1)								:	
													r Rule 23.1(b)).
	ij,			.1	111.	11.					r Rule 48.3(b)	- :	
	֓֞֞֞֜֞֞֜֞֜֞֜֞֜֞֜֞֜֞֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	ιη⊜ 55.2	any an	guag id/or	55.	ជ <u>ុខ</u> ។ 3).	translation t	iumished fo	or the purpo	oses of intern	ational prelim	inary exam	ination (under Rul
	j			9		ij,				•		•	
. VIV	eu Eu	reg tati	aro: ona	to a	iny Jimi	nac	leotide an o V exam inati	d/or amino ion was car	acid sequ rried out on	i ence disclos I the basis of	ed in the inter the sequence	national ap listing	plication, the
		יין. וי	! !	i'] '					219 004401100	nourg.	
	1	ЮΠ	ain	ed h	tir n	m	ernational	application	in written f	orm.			
	1	1		.1 . '1		11 1		- •		mputer reads	able form.	!	
	•	- 1			11.11	4B	ently to this	-					
	1			j 7	9.57	'II '		-		r readable for		ļ	
	ገ ያ ት	ne ne	sta nte	tem mai	enti	thai Nat	the subsec	quently furr s filed has i	nished writt been furnis	en sequence Hed.	listing does n	ot go beyo	nd the disclosure in
	1	G ' ''		1	11 4 11	11:5	-				le form is iden	tical to the	written sequence
	1	Stin	g h	as l	bei	fur	nished.			L	wiiii 10 14011		

4. The amendments have resulted in the cancellation of.

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/GB00/02387

ne description, pages:

the claims. Nos.:

in this report has been established as if (some of) the amendments had not been made, since they have been considered to go beyond the disclosure as filed (Rule 70.2(c)):

(Ally replacement sheet containing such amendments must be referred to under item 1 and annexed to this report.)

6. Additional observations, if necessary:

Y. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

Statement

Novelty (N) Yes: Claims 1-8

No: Claims

Inventive step (IS) Yes: Claims 1-8

No: Claims

industrial applicability (IA) Yes: Claims 1-8

No: Claims

2. Citations and explanations see separate sheet

TERNATIONAL PRELIMINARY EXAMINATION REPORT - SEPARATE SHEET

International application No. PCT/GB00/02387

Re item V

Reasoned statement under Rule 66.2(a)(ii) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

Reference is made to the following document: D1 EP-A-0884271

Article 33(2) famili (3) PCT

lene of the cited prior art documents discloses all the features of the independent claims and 2. Claims 1-8 are therefore novel and fulfill the requirements of Article 33(2) PCT.

. 15 cpnderned with a methanol-reformer which contains a catalyst and is used to supply to afue cet 能 g. claim 1 of D1 and abstract). D1 discusses various ways to reactivate the catalyst whilst it is being used which include:

reducing the finel load; raising the temperature; increasing the air λ value to increase the O, excess

(see sel. 3, 1/10/26, claim 1, col. 5, l. 1st paragraph and col. 6, l. 33-36).

語句 描译 catalysitio be regenerated the reforming reaction is interrupted according to D1, claim. 1. 11 docts not specify that fuel, air and steam continue to be passed through a reforming catalyst during the regeneration phase.

the examples of the application in which fuel, air and steam continue to be passed through a returning catalyst whilst regeneration occurs, either using air pulses or raised temperature, the Applicant has shown that he can maintain a hydrogen concentration of above 25% 耕roughout the operation (example 2, Fig. 1b, example 3). This effect can also be achieved by increasing the steam feed rate (example 5) or by inhibiting deactivation of the catalyst either using an additive (example 4) or extra air (example 6).

Modulating the fivel feed-rate would also alter the oxygen to carbon ratio and be expected ito have a similar effect as modifying the air feed-rate.

Since the methods of claims 1 and 2 are not derivable directly from D1 and the technical ieffects thereby achieved are not foreseeable from D1 an inventive step is accorded.

Chims 1-8 tuffithe requirements of Article 33(3) PCT.

GB00023

0/019903 21 DEC 2001 531 Rec'd PC

2

Because the performance of the reformer has a direct effect on the power output from the fuel cell, deactivation of the catalyst cannot be allowed to proceed unchecked. However the process of power generation cannot be stopped to allow regeneration of the reformer, when its performance falls below an acceptable level. The regeneration of such catalytic reactors will st they are in use is the problem this invention sets out to solve.

the many instances, the performance of the catalyst declines while it is being used. Although the rate of decline can usually be reduced by changing the operating conditions it is often accompanied by a drop in hydrogen and power output. One inequently practised method for restoring the performance of a fuel processor is to replace the catalyst periodically. Another method involves regenerating the catalyst while it remains in the processor, using a prescribed regeneration procedure (eg see JR Roseup Wielsen in Catalysis Today, Vol 37, 1997, p 225-232). However, both these require that the process of hydrogen-generation is interrupted in order to restore performance!

Thus in a first aspect, the present invention provides a method for regenerating a catalytic first processor, while it is being used to supply hydrogen to a fuel cell, comprising any one or more of the steps of:

continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted.

continuing to pass fuel, air and steam through a reforming catalyst and inodulating the air and/or steam feed rate.

continuing to pass, air, fuel and steam through a reforming catalyst and to ulating the feed-rate of the fuel.

continuing to pass fuel, air and steam through a reforming catalyst wherein an exygenate additive is added to the feed.

hand maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

MMENDED SHEET

3

The reforming catalyst may be one component of the catalytic fuel processor, or it may be the only component. Where loss of performance is being caused by a particular component in a complex fuel mixture, the regeneration method can be targeted at the specific deactivating effect.

Thus in a second aspect, the present invention provides a method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:

catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,

continuing to pass fuel, air and steam through a reforming catalyst and

continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel,

continuing to pass fuel, air and steam through a reforming catalyst wherein an oxygenate is added to the feed.

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

Again, the reforming catalyst may be one component or the only component of the catalytic fuel processor. In a preferred embodiment of the invention water is temporarily added to the feed. ('Feed' is a term used to describe the reactant mixture supplied to the fuel processor.) The water will be rapidly converted into steam in the catalyst. The water (steam) to carbon ratio in the feed may readily be adjusted in order to permit targeting of the regeneration method to a specific deactivating event. Adjusted for the purposes of this invention means increased or decreased. The oxygen to carbon ratio may also be adjusted for the same purpose, by adding or changing the feed-rate of air, or by changing the feed-rate of the fuel.

GB00023

5

Figure to shows the effect of air pulses on the composition of dry reformate during reforming of dodecane (as described in Example 2).

In both cases % in reformate is on the vertical axis and time/hours is on the horizontal

represents Hydrogen, ■ represents carbon dioxide, and ▲ represents carbon monoxide.

Figure 2s shows the composition of dry reformate as a function of time during the reforming of toluene under the two conditions described in example 5.

rigure 25 shows the effect of increasing the air feed-rate on the composition of dry reformate, during reforming of toluene (as described in example 6).

in both eases the left hand side of the vertical axis represents % in reformate and the right hand side of the vertical axis represents temperature in °C. The horizontal axis represents time/hours; represents hydrogen, represents carbon dioxide, A represents carbon monoxide and x represents temperature.

will be appreciated that many variations can be made to the invention herein described without departing from the present inventive concept.

Example 1 (Aliphatic fuel; no regeneration)

bed (0.2 g) of reforming catalyst was packed into a tubular quartz reactor, which was positioned at the centre of a furnace. A mixture of dodecane vapour (produced by vaporising the liquid at a rate of 4 cm³ hour-1), air (200 cm³ min-1) and steam (produced by vaporising water at a rate of 4 cm³ hour-1) was passed through the catalyst best, which was heated by the furnace. The furnace temperature was maintained at 50000. The product stream (ie the reformate) passed through a drier before entering a gas chiromotagraph, which was used to analyse for hydrogen. The initial concentration of hydrogen in the dry reformate was just above 25%, but declined to 20% within an hour, and to 15% within 4 hours. See Figure 1a for results.

25'

8

CLAIMS

. 7 O

)::

- A method for regenerating a catalytic fuel processor, while it is being used to supply hydrogen to a fuel cell, comprising any one or more of the steps of:
 - continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,
 - continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,
 - continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.
 - continuing to pass fuel, air and steam through a reforming catalyst wherein an oxygenate is added to the feed.

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

- 2. A method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:
 - continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,
 - continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,
 - continuing to pass air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.
 - continuing to pass fuel, air and steam through a reforming catalyst wherein an exygenate is added to the feed.

AMENDED SHEET

PFC1465

9

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

- A method according to either claim 1 or claim 2, whereby water is temporarily added to the fuel.
- a method according to claims 1 or 2 in which air is temporarily added to the
- A method according to claims 1 or 2 in which an oxygenate is added to the feed.
 - 6 A method according to claim 5 in which the oxygenate is MTBE (methyl-ter butylether).
 - A method according to claims 1 or 2 in which the catalyst bed temperature is raised temporarily by an external heat source.
 - A method according to claims 1 or 2 in which the temperature of one or more of the reaction feeds is raised temporarily.

14

30

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 4 January 2001 (04.01.2001)

PCT

(10) International Publication Number WO 01/00524 A1

(51) International Patent Classification⁷: 3/32, B01J 38/06, 38/04

C01B 3/40,

(21) International Application Number: PCT/GB00/02387

(22) International Filing Date: 20 June 2000 (20.06.2000)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data: 9914662.3

24 June 1999 (24.06.1999) GB

- (71) Applicant (for all designated States except US): JOHN-SON MATTHEY PUBLIC LIMITED COMPANY [GB/GB]; 2-4 Cockspur Street, Trafalgar Square, London SW1Y 5BQ (GB).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): BURCH, Robert [GB/GB]; 34 Belfast Road, Antrim RT41 1PB (GB). GOLUNSKI, Stanislaw, Edmund [GB/GB]; 2 Widmore Lane, Sonning Common, Reading RG4 9RR (GB). SOUTHWARD, Barry, William, Luke [GB/GB]; 19 Ardmore Avenue, Ormeau, Belfast BT7 3HD (GB).

WAILS, David [GB/GB]; 4 Whitecotes Park, Chesterfield, Derbyshire S40 3RT (GB).

- (74) Agent: WISHART, Ian, Carmichael; Johnson Matthey Technology Centre, Blounts Court, Sonning Common, Reading RG4 9NH (GB).
- (81) Designated States (national): AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

Published:

With international search report.

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

1/00524 A1

(54) Title: PROCESS FOR THE REGENERATION OF REFORMING CATALYSTS

(57) Abstract: The present invention relates to methods for the regeneration of catalytic reactors. In particular it relates to methods for regenerating a fuel-processing catalyst whilst it is still being used to supply hydrogen to a fuel cell. The temperature of the catalyst may be adjusted, the air, steam or fuel feed rate may be adjusted. Alternatively, additives may be added to the feed.

PROCESS FOR THE REGENERATION OF REFORMING CATALYSTS

1

The present invention relates to methods for the regeneration of catalytic reactors.

5

10

15

20

Catalytic converters are frequently used in internal combustion engines in order to meet the various regulations concerning the levels of pollutants in exhaust gases. A three-way catalyst is a common form of converter used. This has three main duties, namely, the oxidation of CO, the oxidation of unburnt hydrocarbons (HC's) and the reduction of NOx to N2. Such catalysts require careful engine management to ensure that the engine operates at or close to stoichiometric conditions, that is fuel/air lambda=1. Growing awareness of the need to conserve the earth's resources and increasingly stringent legislation have recently prompted a search for cleaner and more efficient alternatives to the internal combustion engine. One of the most promising of these is the combination of an electric motor and a fuel cell. However, the latter requires a source of hydrogen, for which there is no supply and distribution infrastructure comparable to that for liquid fuels.

The use of fuel cells is not just limited to vehicle applications. Emergent markets include domestic co-generation of heat and power, and power generation in remote locations. Again, the availability of hydrogen is a key issue, with natural gas often being the preferred fuel for domestic systems, and liquid hydrocarbons being more transportable to remote locations.

25

The problems of hydrogen supply and distribution can be overcome by generating it within the fuel-cell system. A hydrogen-rich gas stream, commonly known as 'reformate', can be produced by catalytically converting organic fuels (such as gasoline, natural gas or alcohol). The process, which is referred to as 'reforming' or 'fuel-processing', can occur by a number of different reaction mechanisms:

- 30 dissociation (splitting of the fuel molecules),
 - steam reforming (reaction of the fuel with H₂O),
 - partial oxidation (reaction of the fuel with O2, usually supplied as air)
 - combinations of the above reactions.

Because the performance of the reformer has a direct effect on the power output from the fuel cell, deactivation of the catalyst cannot be allowed to proceed unchecked. However the process of power generation cannot be stopped to allow regeneration of the reformer, when its performance falls below an acceptable level. The regeneration of such catalytic reactors whilst they are in use is the problem this invention sets out to solve.

2

In many instances, the performance of the catalyst declines while it is being used. Although the rate of decline can usually be reduced by changing the operating conditions, it is often accompanied by a drop in hydrogen and power output. One frequently practised method for restoring the performance of a fuel processor is to replace the catalyst periodically. Another method involves regenerating the catalyst while it remains in the processor, using a prescribed regeneration procedure (eg see JR Rostrup-Nielsen in Catalysis Today, Vol 37, 1997, p 225-232). However, both these require that the process of hydrogen-generation is interrupted in order to restore performance.

Thus in a first aspect, the present invention provides a method for regenerating a catalytic fuel processor, while it is being used to supply hydrogen to a fuel cell, comprising any one or more of the steps of:

20

25

15

5

ΤÖ

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted.

- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate.

- continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.
- continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

30

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

3

The reforming catalyst may be one component of the catalytic fuel processor, or it may be the only component. Where loss of performance is being caused by a particular component in a complex fuel mixture, the regeneration method can be targeted at the specific deactivating effect.

5

Thus in a second aspect, the present invention provides a method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:

10

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,
- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,

15

25

30

- continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel,
- continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

Again, the reforming catalyst may be one component or the only component of the catalytic fuel processor. In a preferred embodiment of the invention water is temporarily added to the feed. ('Feed' is a term used to describe the reactant mixture supplied to the fuel processor.) The water will be rapidly converted into steam in the catalyst. The water (steam) to carbon ratio in the feed may readily be adjusted in order to permit targeting of the regeneration method to a specific deactivating event. Adjusted for the purposes of this invention means increased or decreased. The oxygen to carbon ratio may also be adjusted for the same purpose, by adding or changing the feed-rate of air, or by changing the feed-rate of the fuel.

The catalyst may contain one or more base metals, which may include copper (often used for reforming methanol) and nickel (used for reforming natural gas and

higher hydrocarbons). Alternatively, the catalyst may contain one or more precious

metals, which may include gold, platinum, palladium, iridium, silver, rhodium and

ruthenium. Furthermore, it may contain both base metal(s) and precious metal(s).

The catalyst may also contain refractory materials, such as ceramics, metal oxides,

perovskites, metal carbides and metal sulphides.

10

15

20

25

30

Additives may be added to the feed for various purposes, including acceleration of start-up of a fuel-processor, and the prevention or inhibition of its deactivation. Alternatively, a fuel to which additives have been added during manufacture can be used. In a preferred embodiment of the invention, the additive is an oxygenate, and in an especially preferred embodiment the oxygenate is MTBE (methyl-tert-butylether).

In order to facilitate the regeneration procedure, the temperature of the catalyst bed may be raised temporarily by an external energy source. Within a fuel cell system, the external energy source may be an electrical heater, or a burner (which combusts either some of the fuel or some of the hydrogen produced). In addition, or alternatively, the temperature of one or more of the feed components may be raised temporarily, again by an external heat source.

Among the most common causes of deactivation of a fuel-processor is the retention of carbon or sulphur species by the catalyst. Therefore, in a further embodiment of the invention, catalyst regeneration may occur by removal of the carbon or sulphur species. The carbon species originate from the fuel molecules, and indicate the occurrence of undesired side reactions. The sulphur species originate either from indigenous contaminants in the fuel or from compounds deliberately added to the fuel (such as the odourants used to give natural gas its recognisable smell).

The present invention will now be described by way of the following examples in which:

Figure 1a: shows the composition of dry reformate as a function of time during reforming of dodecane (under conditions described in Example 1).

Figure 1b shows the effect of air pulses on the composition of dry reformate during reforming of dodecane (as described in Example 2).

In both cases % in reformate is on the vertical axis and time/hours is on the horizontal axis.

5 ◆ represents Hydrogen, ■ represents carbon monoxide, and ▲ represents carbon monoxide.

Figure 2a shows the composition of dry reformate as a function of time during the reforming of toluene under the two conditions described in example 5.

Figure 2b shows the effect of increasing the air feed-rate on the composition of dry reformate, during reforming of toluene (as described in example 6).

In both cases the left hand side of the vertical axis represents % in reformate and the right hand side of the vertical axis represents temperature in °C. The horizontal axis represents time/hours; ◆ represents hydrogen, ■ represents carbon dioxide,

▲ represents carbon monoxide and x represents temperature.

15

20

25

30

It will be appreciated that many variations can be made to the invention herein described without departing from the present inventive concept.

Example 1 (Aliphatic fuel; no regeneration)

A bed (0.2 g) of reforming catalyst was packed into a tubular quartz reactor, which was positioned at the centre of a furnace. A mixture of dodecane vapour (produced by vaporising the liquid at a rate of 4 cm³ hour-¹), air (200 cm³ min-¹) and steam (produced by vaporising water at a rate of 4 cm³ hour-¹) was passed through the catalyst bed, which was heated by the furnace. The furnace temperature was maintained at 500°C. The product stream (ie the reformate) passed through a drier before entering a gas chromotagraph, which was used to analyse for hydrogen. The initial concentration of hydrogen in the dry reformate was just above 25%, but declined to 20% within an hour, and to 15% within 4 hours. See Figure 1a for results.

Example 2 (Regeneration by air pulses)

The test procedure described in Example 1 was repeated with a fresh charge of catalyst, except that every 10 minutes the air feed-rate was increased to 350 cm³ min-¹ for 30 seconds. Apart from the duration of the extra air pulses, the concentration of hydrogen in the dry reformate remained above 25% during 3 hours of testing. See Figure 1b for results.

Example 3 (Regeneration by temperature excursions)

10

15

20

25

30

5

The test procedure described in Example 1 was repeated with a fresh charge of catalyst. Whenever the hydrogen concentration in the dry reformate dropped below 25%, it could be restored by raising the furnace temperature to 600°C for 1 minute.

Example 4 (Inhibition of de-activation by MTBE addition)

The test procedure described in Example 1 was repeated with a fresh charge of catalyst, except that 10% (by volume) methyl-tert-butylether was added to the dodecane. The hydrogen concentration in the dry reformate remained above 25% throughout 5 hours of testing.

Example 5 (Aromatic fuel; no regeneration)

A bed (0.2 g) of reforming catalyst was packed into a tubular quartz reactor, which was positioned at the centre of a furnace. A mixture of toluene vapour (produced by vaporising the liquid at a rate of 4 cm³ hour-¹), air (175 cm³ min-¹) and steam (produced by vaporising water at a rate of 4 cm³ hour-¹) was passed through the catalyst bed, which was heated by the furnace. The furnace temperature was maintained at 500°C. The initial concentration of hydrogen in the dry reformate was 33%, but declined to 25% within 3 hours. When the catalyst was replaced by a fresh charge and the feed-rate of steam was doubled (by increasing the rate of vaporising water to 8 cm³

hour-1), the initial concentration of hydrogen in the dry reformate was 37%. Within 3 hours, the hydrogen concentration had declined to 30%. See Figure 2a for results.

Example 6 (Prevention of de-activation by extra air)

5

10

The test procedure described in Example 5 was repeated with a fresh charge of catalyst, except that the air feed-rate was increased to 200 cm³ min-¹. The initial concentration of hydrogen in the dry reformate was 32%. The concentration remained unchanged during 3 hours of testing. See Figure 2b for results.

CLAIMS

1. A method for regenerating a catalytic fuel processor, while it is being used to supply hydrogen to a fuel cell, comprising any one or more of the steps of:

5

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,

10

- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,

10

- continuing to pass, air, fuel and steam through a reforming catalyst and modulating the feed-rate of the fuel.
- continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

15

- and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.
- 2. A method for preventing or retarding the de-activation of a catalytic fuel processor while it is being used to supply hydrogen to a fuel cell comprising any one or more of the steps of:

25

30

- continuing to pass fuel, air and steam through a reforming catalyst whilst the catalyst is heated by an external heat source such that the temperature of the catalyst may be adjusted,

- continuing to pass fu

- continuing to pass fuel, air and steam through a reforming catalyst and modulating the air and/or steam feed rate,

continuing to pass air, fuel and steam through a reforming catalyst and

modulating the feed-rate of the fuel.
- continuing to pass fuel, air and steam

continuing to pass fuel, air and steam through a reforming catalyst wherein additive is added to the feed.

and maintaining the hydrogen concentration in dry reformate above 25% throughout the operation of the processor.

- 3. A method according to either claim 1 or claim 2, whereby water is temporarily added to the fuel.
 - 4. A method according to claims 1 or 2 in which air is temporarily added to the feed.
- 10 5. A method according to claims 1 or 2 in which an additive is added to the feed.
 - 6. A method according to claim 5 in which the additive is an oxygenate.
- 7. A method according to claim 6 in which the oxygenate is MTBE 15 (methyl-tert-butylether).
 - 8. A method according to claims 1 or 2 in which the catalyst bed temperature is raised temporarily by an external heat source.
- 9. A method according to claims 1 or 2 in which the temperature of one or more of the reactant feeds is raised temporarily.

1/2

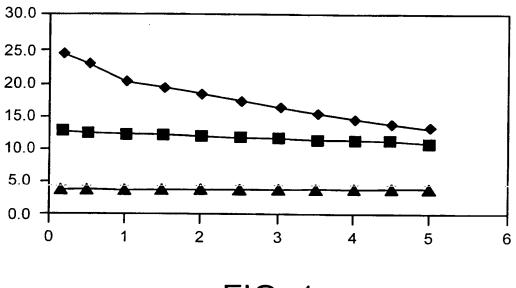


FIG. 1a

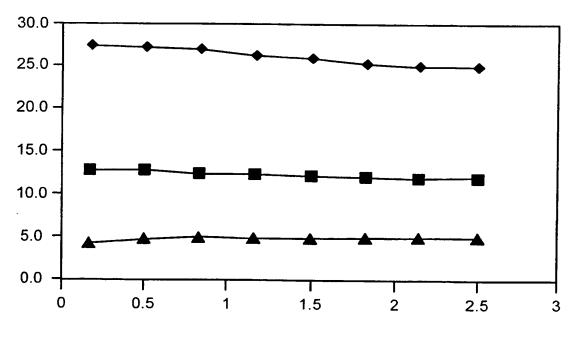
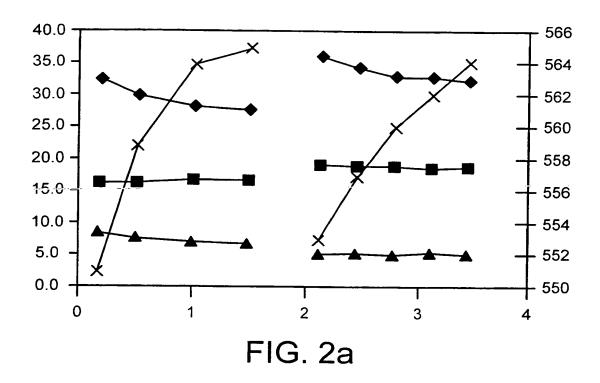
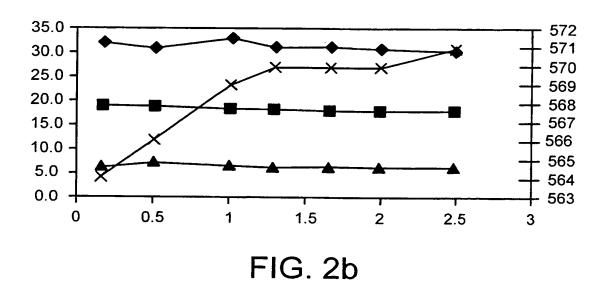


FIG. 1b





A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C01B3/40 C01B3/32

B01J38/06

B01J38/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols) $IPC\ 7\ C01B\ B01J$

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

WPI Data, EPO-Internal, CHEM ABS Data, PAJ, INSPEC, COMPENDEX, API Data

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 884 271 A (DBB FUEL CELL ENGINES GMBH) 16 December 1998 (1998-12-16) the whole document	1,2,5,8,
A	US 4 089 941 A (VILLEMIN BERNARD) 16 May 1978 (1978-05-16) claim 7	1,2
A	US 5 075 268 A (KURASHIGE MITSUHIKO ET AL) 24 December 1991 (1991-12-24) the whole document	1,2
A	US 4 083 799 A (ESTES JOHN HAROLD ET AL) 11 April 1978 (1978-04-11) column 6, line 14 - line 18 -/	1,2
•		·

Further documents are listed in the continuation of box C.	Patent family members are listed in annex.
Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention of carnot be considered novel or carnot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention carnot be considered to involve an inventive step when the document is carnot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "&" document member of the same patent family
Date of the actual completion of the international search	Date of mailing of the International search report
25 September 2000	02/10/2000
Name and mailing address of the ISA	Authorized officer
European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijawijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Van der Poel, W

1



inter mai Application No PCT/GB 00/02387

(Continu	etion) DOCUMENTS CONSIDERED TO BE RELEVANT	PCT/GB 0	0/02387
tegory *	Citation of document, with indication, where appropriate, of the relevant passages		
	on the relevant passages		Relevant to claim No.
A	PATENT ABSTRACTS OF JAPAN vol. 1999, no. 08, 30 June 1999 (1999-06-30) & JP 11 079702 A (TOYOTA CENTRAL RES &DEV LAB INC), 23 March 1999 (1999-03-23) abstract		1,2,4
1	DE 24 03 701 A (BASF AG) 7 August 1975 (1975-08-07) claims		1,2,6
	US 5 043 518 A (MICHAELSON ROBERT C ET AL) 27 August 1991 (1991-08-27) column 2, line 40 -column 4, line 21		6,7
	•		
		·	

INTERNATIONAL SEARCH REPORT

.nformation on patent family members

onal Application No PCT/GB 00/02387

Publication Patent family Patent document **Publication** member(s) cited in search report date date 19725007 C 18-03-1999 EP 0884271 Α 16-12-1998 DE FR 20-05-1977 US 4089941 A 16-05-1978 2328656 A 28-04-1977 DE 2645522 A DK 476176 A,B, 23-04-1977 GB 1511789 A 24-05-1978 JP 52052191 A 26-04-1977 24-12-1991 JP 1846686 C 07-06-1994 US 5075268 Α JP 4200640 A 21-07-1992 JP 12-11-1993 5081303 B US 4083799 Α 11-04-1978 NONE JP 11079702 Α 23-03-1999 NONE DE 2403701 Α 07-08-1975 NONE 27-08-1991 CA 01-02-1991 US 5043518 Α 2064040 A 15-02-1995 EP 0637994 A 24-12-1992 JP 4507368 T WO 9101804 A 21-02-1991 29-06-1993 US 5223464 A